

Nuclear Spin-Independent Effects of Parity NonConservation in Molecule of Hydrogen

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Theoretical study of discrete symmetry (e.g. spatial parity \mathcal{P}) violation effects in atomic and molecular systems plays a key role in developing theories and models of fundamental interaction physics. In all atomic experiments only the \mathcal{P} -odd electron-nucleus (e -N) interaction was observed since it is enhanced in heavy atoms, while the \mathcal{P} -odd electron-electron (e - e) interaction is negligible [1]. In 1977 it was proposed to observe the effect of optical rotation on oxygen molecule [2]. According to very rough estimates in this case the \mathcal{P} -odd e - e interaction effect should prevail over the \mathcal{P} -odd e -N one. The idea was that all electrons of a molecule involved in formation of the chemical bond are concentrated in the region between the nuclei, so there should not arise additional smallness in the matrix element of the \mathcal{P} -odd e - e interaction, which is present in atoms for that effect. It follows from our recent calculations via coupled cluster method [3] that for O_2 the \mathcal{P} -odd e - e interaction is suppressed compared to the \mathcal{P} -odd e -N one and, in principle, cannot be separated out in this case. Note also that this suppression in O_2 is less than the one in an atom with the same nuclear charge.

This contribution is devoted to the description and the calculation of the \mathcal{P} -odd effects in diatomic homonuclear molecule of parahydrogen H_2 . For this purpose the M1 transition between the states with the same rotational number of the vibrational $v = 1 \leftarrow 0$ band in the H_2 ground electronic $^1\Sigma_g^+$ state [4] is considered. It is shown that in this case the effects of the \mathcal{P} -odd e - e and e -N spin-independent interactions are of the same order of magnitude. H_2 molecule is therefore the first example of atomic system where the e - e PNC interaction can be directly observed. Since the constants of the e -N PNC interaction were already accurately measured in atomic experiments, the e - e PNC interaction constant also can be extracted from these experiments. In all other atoms and molecules the e - e PNC interaction is usually deeply screened by the e -N interaction. Then the evaluations of the PNC effect in the parahydrogen molecule H_2 may acquire another important sense. The parahydrogen molecule H_2 is the unique atomic system also because in this molecule the e - e and e -N weak interaction constants are the same. Then a moment's consideration shows that the PNC experiments with H_2 molecule may become a source of the most accurate values for the Weinberg's angle.

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